

Novel Synthetic Strategy to Chiral Alkylated Lactams Employing C₂-Symmetry

Hidemi Yoda,* Hidekazu Kitayama, Wataru Yamada,
Takao Katagiri, and Kunihiro Takabe*

Department of Applied Chemistry, Faculty of Engineering,
Shizuoka University, Hamamatsu 432, Japan

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Abstract: Reductive deoxygenation of quarternary α -hydroxy lactams, readily prepared from alkylation of C₂-symmetrical imides, with Et₃SiH in the presence of Lewis acid displayed an extremely high stereoselectivity to provide the corresponding lactams. Stereochemistry of the new stereogenic center generated has been confirmed unambiguously to be S, resulting from an unprecedented *trans*-selective reaction, by the transformation into the known lactam.

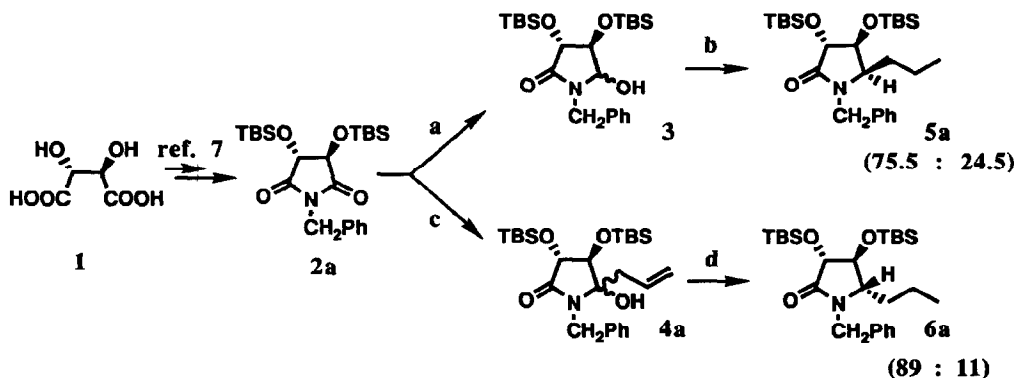
Owing to their useful structural features in the synthesis of chiral substances possessing a wide range of biological activity, asymmetric reactions employing reagents with a C₂-axis of symmetry are of great interest.¹ Over the last two decades they have led to a rapid growing speciality in the general field of synthetic organic chemistry. Therefore, a number of efficient techniques and synthetic strategies have appeared for the construction of those compounds in optically pure form² and further exploitation of much more convenient methods is subject to continual refinement.

On the other hand, the synthetic utility of *N*-acyliminium ions obtained from the partial reduction of cyclic imides has been demonstrated in the preparation of nitrogen-containing natural products such as alkaloids.³ In addition, very recently moderate *cis*-selective intermolecular alkylation of an α -alkoxy *N*-acyliminium ion was first reported⁴ as part of a radical cyclization methodology.⁵ In spite of the impressive behavior of such intermediates, little, if any, effort has been made for the utilization of the quarternary carbon center present in the α -hydroxy lactams (4).⁶

As part of our recent continuing investigations designed to extend the employment of C₂-symmetrical imide (2), a diastereomer differentiated reaction⁷ and the application to the total synthesis of natural cerulenin, antibiotic⁸ have been developed from this laboratory. The central purpose of the present communication is to detail the unprecedented results that *trans*-selective (with respect to the C-4 substituent) deoxygenation of the quarternary α -hydroxy lactams (4)⁹ could be accomplished exclusively.

In order to introduce an alkyl group stereoselectively, homochiral imide (2a) prepared in one pot from L-

tartaric acid was first reduced with NaBH_4 to give a mixture of two stereoisomeric hydroxy lactams (**3**).¹⁰ As shown in Scheme 1, acid-induced alkylation of **3** proceeding via an *N*-acyliminium intermediate^{4,5} upon treatment with allyltrimethylsilane in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ resulted in the moderate *cis*-selective formation of **5a**¹¹ at 0 °C (**5a** : **6a** = 75.5 : 24.5 determined by HPLC¹²). Next, we investigated the utilization of the alkylated quarternary α -hydroxy lactam (**4a**) prepared by the Grignard addition to **2a** according to our method.⁷ When **4a** was treated with Et_3SiH in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ at 0 °C, it reversely afforded deoxygenated lactam **6a** cleanly and *trans*-selectively in the ratio of 89 : 11.¹¹



Scheme 1. Reagents and conditions: (a) NaBH_4 , MeOH, -15 °C; 91%; (b) **1**, allyltrimethylsilane, $\text{BF}_3 \cdot \text{OEt}_2$, CH_2Cl_2 , 0 °C; 96%; **2**, H_2 , Pd/C, EtOH; 100%; (c) allylmagnesium bromide, THF, -78 - 0 °C; 74%; (d) Et_3SiH , $\text{BF}_3 \cdot \text{OEt}_2$, CH_2Cl_2 , 0 °C; 87%.

With the above unusual stereoselective results in hand, we further examined the reactions with several types of **4** under similar conditions but at low temperature (-78 °C) in order to enhance the selectivity. The details are summarized in Table 1. The reaction proceeded in high yield in each case. Increasing the steric bulkiness of the silyl-protecting groups as well as lowering the reaction temperature led to an increase of the diastereoselectivity in the reaction (Entries 3:4 and 5:7), however, the products were still obtained with high selectivity even with TBS groups (Entries 1,2). A change from the benzyl substituent on the nitrogen atom to a smaller methyl group enhanced the selectivity a little (Entries 3:5 and 4:7). In the case of the reactions with a benzyl side chain, complete diastereodifferentiation was observed (Entries 6,9, and 10).

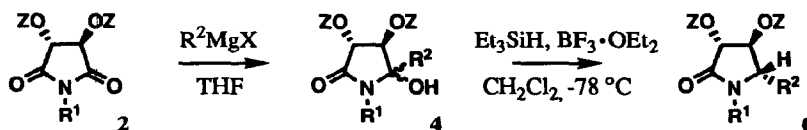
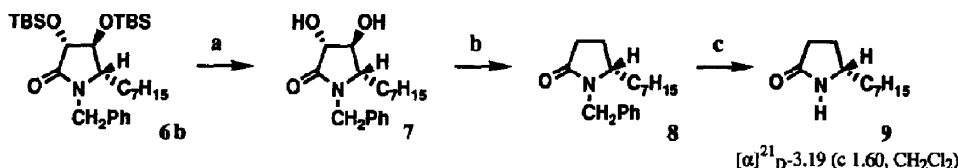


Table 1 Reductive Deoxygenation of quaternary α -hydroxy lactams (**4**)^a

Entry	R ¹	Z	R ²	Yield ^b) of 4 (%)	Yield ^b) of 6 (%)	Ratio ^c) of 6 (5 <i>S</i>) : (5 <i>R</i>)
1	PhCH ₂	TBS	n-C ₃ H ₇	98	78 (6a)	95.1 : 4.9
2	PhCH ₂	TBS	n-C ₇ H ₁₅	95	91 (6b)	94.7 : 5.3
3	PhCH ₂	TBS	n-C ₈ H ₁₇	99	76 (6c)	95.5 : 4.5
4	PhCH ₂	TIPS	n-C ₈ H ₁₇	100	98 (6d)	97.2 : 2.8
5	CH ₃	TBS	n-C ₈ H ₁₇	97	96 (6e)	97.2 : 2.8
6	CH ₃	TBS	PhCH ₂	91	83 (6f)	> 99 : 1
7	CH ₃	TIPS	n-C ₈ H ₁₇	98	95 (6g)	98.5 : 1.5
8	CH ₃	TIPS	n-C ₃ H ₇	95	85 (6h)	> 99 : 1
9	CH ₃	TIPS	PhCH ₂	91	70 (6i)	> 99 : 1
10	CH ₃	TIPS	<i>p</i> -MeOPhCH ₂	-	55 ^d) (6j)	> 99 : 1

a) 5-10 equiv. of Et₃SiH and BF₃·OEt₂ were used. b) Isolated yield. c) Determined by HPLC (Daicel Chiralpak AS) and ¹³C NMR. d) Based on **2**.

The stereochemistry of the newly created stereogenic center of **6** was proven by transformation into known lactam **9**, as shown in Scheme 2. Compound **6b** (Entry 2) was submitted to sequential desilylation and dehydration with triiodoimidazole in the presence of Zn^{7a,13} followed by hydrogenation, leading to the saturated lactam (**8**). Finally, **8** was subjected to debenzylation to produce **9**, which could be assigned as *S* form in comparison of its specific rotation with that reported.^{9b}



Scheme 2. Reagents and conditions: (a) concd. HCl, MeOH; 95%; (b) 1, triiodoimidazole, imidazole, Ph₃P, Zn, toluene, 70 °C; 2, Mg, MeOH; 49% (2 steps); (c) Na, NH₃, -78 °C; 97%.

Although the reason why such an unusual *trans*-selective deoxygenation reaction was observed is not clarified at present, an efficient and general method to construct chiral alkylated lactams has been established by using C₂-symmetrical imide as a template. This strategy provides a new synthetic opportunity for the synthesis of natural indolizidine alkaloids. The following paper describes our results concerning the total synthesis of biologically active lentiginosine.

References and notes

- See, for example: Katsuki, T.; Sharpless, K. B. *J. Am. Chem. Soc.* **1980**, *102*, 5974; Noyori, R. *Pure Appl. Chem.* **1981**, *53*, 2315; Noyori, R. "Advances in Asymmetric Synthesis and Optical Resolution" ed by Otsuka, S.; Mukaiyama, T. Kagaku-doizin **1982**, Chapter 5; Katsuki, T.; Yamaguchi, M. *Yuki Gosei Kagaku Kyokaiishi* **1986**, *44*, 532; Sakamoto, A.; Yamamoto, Y.; Oda, J. *J. Am. Chem. Soc.* **1987**, *109*, 7188; Maruoka, K.; Itoh, T.; Shirasaka, T.; Yamamoto, H. *J. Am. Chem. Soc.* **1988**, *110*, 310; Yamamoto, Y.; Sakamoto, A.; Nishioka, T.; Oda, J. *J. Org. Chem.* **1991**, *56*, 1112.
- Saigo, K.; Kubota, N.; Takebayashi, S.; Hasegawa, M. *Bull. Chem. Soc. Jpn.*, **1986**, *59*, 931; Schlessinger, R. H.; Iwanowicz, E. J. *Tetrahedron Lett.*, **1987**, *28*, 2083; Takano, S.; Moriya, M.; Iwabuchi, Y.; Ogasawara, K. *ibid.*, **1989**, *30*, 3805; Short, R. P.; Kennedy, R. M.; Masamune, S. *J. Org. Chem.* **1989**, *54*, 1755; Watson Jr., H. A.; O'Neill, B. T. *ibid.*, **1990**, *55*, 2950.
- Bienz, S.; Busacca, C.; Meyers, A. I. *J. Am. Chem. Soc.* **1989**, *111*, 1905; Klaver, W. J.; Hiemstra, H.; Speckamp, W. N. *ibid.*, **1989**, *111*, 2588 and references cited therein; Miller, S. A.; Chamberlin, A. R. *J. Org. Chem.* **1989**, *54*, 2502; Miller, S. A.; Chamberlin, A. R. *J. Am. Chem. Soc.* **1990**, *112*, 8100; Gelas-Mjalhe, Y.; Gramain, J.-C.; Louvet, A.; Remuson, R. *Tetrahedron Lett.*, **1992**, *33*, 73.
- Bernardi, A.; Micheli, F.; Potenza, D.; Scolastico, C.; Villa, R. *Tetrahedron Lett.*, **1990**, *31*, 4949.
- Koot, W.-J.; Ginkel, R.; Kranenburg, M.; Hiemstra, H.; Louwrier, S.; Moolenaar, M. J.; Speckamp, W. N. *Tetrahedron Lett.*, **1991**, *32*, 401 and references cited therein.
- Yoda, H.; Morishita, H.; Kudo, M.; Katagiri, T.; Takabe, K. *Chem. Express*, **1989**, *4*, 515.
- (a) Yoda, H.; Shirakawa, K.; Takabe, K. *Chem Lett.* **1991**, 489. (b) Yoda, H.; Shirakawa, K.; Takabe, K. *Tetrahedron Lett.*, **1991**, *32*, 3401.
- Yoda, H.; Katagiri, T.; Takabe, K. *Tetrahedron Lett.*, **1991**, *32*, 6771.
- Recently two independent groups reported stereoselective deoxygenation of a quaternary α -hydroxy (or alkoxy) lactam elucidated via an *N*-acyliminium ion: (a) Ohta, T.; Shiokawa, S.; Sakamoto, R.; Nozoe, S. *Tetrahedron Lett.*, **1990**, *31*, 7329; (b) Burgess, L. E.; Meyers, A. I. *J. Org. Chem.* **1992**, *57*, 1656, in which (*R*)-**9** was shown to have $[\alpha]_D^{26} +9.0$ (c 2.0, CH₂Cl₂). The chiral center of (*S*)-**9** thus obtained seems to racemize during the dehydration process.
- The ratio of the two stereoisomers was not determined.
- ¹H NMR data (CDCl₃, 90 MHz) for **5** and **6a**. **5**: δ 0.00 (3H, s), 0.10 (3H, s), 0.17 (3H, s), 0.23 (3H, s), 0.76-1.56 (7H, m), 0.89 (9H, s), 0.95 (9H, s), 3.39 (1H, dt, $J_{4,5}, J_{5,6} = 6.2, 6.2$ Hz), 3.81-4.25 (3H, m), 4.95 (1H, d, $J = 15.2$ Hz), 7.28 (5H, s). **6a**: δ 0.01 (3H, s), 0.08 (3H, s), 0.19 (3H, s), 0.21 (3H, s), 0.73-1.61 (7H, m), 0.84 (9H, s), 0.93 (9H, s), 3.15 (1H, dt, $J_{4,5}, J_{5,6} = 2.0, 5.9$ Hz), 3.85 (1H, dd, $J = 2.0, 2.0$ Hz), 3.90 (1H, d, $J = 15.4$ Hz), 4.02 (1H, d, $J = 2.0$ Hz), 5.09 (1H, d, $J = 15.4$ Hz), 7.24 (5H, s). Observed chemical shift and vicinal coupling constants ($J_{4,5}$ and $J_{5,6}$) of the other major isomers (**6b** - **6j**)^{4,5} were almost identical with those of **6a**. And the absolute configuration of the generated stereogenic center of **6** was clearly determined as shown in Scheme 2.^{9b}
- HPLC conditions were as follows. Column: Daicel Chiralpak AS, 4.6 x 250mm. Eluent: hexane - 2-propanol (800: 1), 0.7 ml/min. Detection: UV at 220 nm.
- Yamazaki, N.; Kibayashi, C. *J. Am. Chem. Soc.* **1989**, *111*, 1396.